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An efficient, enantioselective synthesis of branched polyhydroxylated pyrrolidines[†]

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Abstract

The enantioselective synthesis of branched polyhydroxylated pyrrolidines from a novel α -methylene bicyclic lactam (4b) is described. © 2000 Elsevier Science Ltd. All rights reserved.

Polyhydroxylated pyrrolidines have generated a great deal of synthetic interest in the past decade due to their structural resemblance to the sugar moiety of monosaccharides.¹ Because of their ability to mimic carbohydrates, they function as potent inhibitors of glycosidases; enzymes that are involved in a wide range of important biological processes.² The discovery that these sugar mimetics display considerable activity against cancer,³ diabetes,¹ and viral infections (anti-HIV behavior)⁴ has led to an enhanced interest in them as potential therapeutic agents. Recently, the syntheses of novel polyhydroxylated piperidines, **1**, and pyrrolidines, (+)-**2**, containing branched skeletons⁵ (Scheme 1) have been reported in an attempt to develop more selective and potent glycosidase inhibitors. Due to our previous success in preparing azasugars from bicyclic lactams⁶ and our continued effort in expanding the versatility of this chiral molecule, we anticipated that the synthesis of branched polyhydroxylated pyrrolidines, **2**, could be achieved in high enantiomeric purity from an appropriately functionalized 5,5-bicyclic lactam (**4b**).

It was envisioned that bicyclic lactam **4b**, possessing an α -methylene substituent, could be accessed from the known lactam **3b**.⁶ The olefin, **4b**, would be subjected to dihydroxylation conditions to afford diol **5b**—hopefully with some degree of *exo–endo* selectivity. Reductive cleavage of **5b**, after diol protection, would complete an efficient synthesis of pyrrolidine **6**. The latter could be further elaborated to azasugars such as **2**, with the option of placing a variety of substituents (R²) on the nitrogen atom.

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[†] This paper is dedicated with respect and admiration to Professor Harry Wasserman on his 80th birthday.



Scheme 1.

The route to **2** and **6** began with the preparation of known bicyclic lactam $3b^6$ from 2,3-dihydrofuran in 47% yield over three steps. Readily available lactam 3a,⁷ containing an angular methyl substituent, was also prepared and employed as a model. Attention was then turned toward the introduction of the α -methylene substituent. Although a variety of methods to prepare α -methylene lactones are known,⁸ all attempts to use these met with failure or limited success. Diez and co-workers⁹ reported similar difficulties in preparing chiral α -methylene 2-piperidones. In an effort to avoid these problems, the enolate anions of both lactams **3a** and **3b** were treated with diethyl phosphorochloridite¹⁰ followed by immediate air oxidation of the resultant reaction products (Scheme 2). In both cases (lactams **7a** and **7b**), the ³¹P NMR spectra of the crude reaction mixtures revealed a 2.5:1 ratio of C–P to O–P bond formation products—the C–P products being a mixture of *exo–endo* diastereomers (3:1). This crude mixture was treated with NaH followed by paraformaldehyde to provide a 43% yield (55% yield based on recovered starting material) of the desired α -methylene bicyclic lactams **4a** and **4b** over the two steps.

Initial experiments to introduce vicinal hydroxyls were performed on lactam 4a, which was treated with OsO₄-NMO to furnish a disappointing 3:1 (NMR) mixture of diastereomers (Table 1). In order to increase the selectivity of the dihydroxylation, quinuclidine was added as a ligand since it was anticipated that the ligand–OsO₄ complex¹¹ would provide increased facial selectivity due to its greater steric bulk as compared to OsO4 alone. The selectivity did increase, providing a 7:1 mixture of diastereomers at the newly formed tertiary alcohol center (diol 5a). Attention was now turned to bicyclic lactam 4b. Treatment of 4b with OsO₄-quinuclidine, however, gave a poor mixture of diastereomers (3:1) as compared to the model system (7:1). This result suggested that the dihydroxylation might be occurring preferentially from the exo-face; the decrease in exo-facial selectivity with lactam 4b perhaps arising from the larger angular benzyloxymethyl substituent shielding the β -face of the lactam. It was felt that the addition of a ligand bulkier than quinuclidine might improve the facial selectivity of the dihydroxylation. Lactam 4b was subsequently treated with AD-mix- α and AD-mix- β , respectively¹¹ providing a greater than 15:1 (NMR) mixture of diastereomers. It is interesting to note that in the reactions using either AD-mix- α or AD-mix- β , the same ratio of diastereomers was observed (15:1) with the same sense of stereochemistry. This strongly suggests that the



Scheme 2.

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Lactam	Ligand	de	Yield (%)
	_	3:1	89
4a	Quinuclidine	7:1	96
4b	Quinuclidine	3:1	93
4b	AD-mix-α	15:1	98
4b	AD-mix-β	15:1	100

increase in the observed facial selectivity is not due to the chirality of these ligands, but is due solely to their steric bulk. The stereochemistry of the dihydroxylation products 5, were tentatively assigned, at this point, as taking place from the *exo*-face.

The two hydroxyl groups were transformed to their benzyl ethers, **8**, and the C–O cleavage step was next addressed to prepare pyrrolidine **9**. From earlier results in our laboratory,¹² it was anticipated that this reaction would proceed with retention of configuration of the angular substituent. Thus, treatment of **8** with DIBAL-H furnished an 88% yield of pyrrolidine **9**; the reductive cleavage of the ring C–O bond occurring with very high diastereoselectivity (no trace of C-2 diastereomers being observed). The 15:1 mixture of diastereomers obtained from the dihydroxylation could also be separated by column chromatography at this point, providing compound **9** as a single diastereomer. The stereochemistry at C-2 was tentatively assigned as

having occurred with retention of configuration. Catalytic hydrogenation $(H_2, 4 \text{ atm}, Pd(OH)_2)$ in the presence of Boc₂O afforded branched pyrrolidine **6**, which was immediately acetylated to provide **10** in 68% yield. It is interesting to note that if Boc₂O is not added to the hydrogenation, only *N*-benzyl deprotection is observed and the *O*-benzyl groups remain intact.¹³ Pyrrolidine **10** provides a convenient entry into branched pyrrolidines, because by simply treating **10** with TFA in CH₂Cl₂, the Boc protecting group can be removed (pyrrolidine **11**), leaving the nitrogen free to react with any number of electrophiles. The ease at which **10** is transformed into *N*-benzylated pyrrolidine **2** illustrates the usefulness of this intermediate.

With an efficient, enantioselective synthesis of pyrrolidine 10 in hand (ten steps from 2,3-dihydrofuran, 12% overall yield), it was necessary to confirm the absolute configuration. Commercially available (S)-(+)-5-(hydroxymethyl)-2-pyrrolidinone 12 was dibenzylated to afford pyrrolidinone 13 in 80% yield (Scheme 3). The latter was olefinated [LDA/(EtO)₂PCl, air, NaH/ $(CH_2O)_n$) to the α -methylene lactam 14 in 65% yield. Dihydroxylation of 14 employing OsO₄-NMO gave a 2.5:1 (NMR) mixture of diastereomers (diol 15). Similar results were obtained using quinuclidine. Surprisingly, addition of AD-mix- α or AD-mix- β decreases the diastereoselectivity, affording a 1:1 ratio of diastereomers. It was assumed that the major product in the OsO_4 -NMO dihydroxylation arose from α -facial selectivity; the *trans* product being favored due to shielding of the β -face by the benzyloxymethyl C-5 substituent. Dibenzylation of the diol moiety followed by a DIBAL-H reduction furnished the tetrabenzyl pyrrolidine 17 in high yield. Hydrogenolyses of the N-benzyl and O-benzyl groups in the presence of Boc₂O afforded 18, which was immediately acetylated to provide pyrrolidine 19 as a 3:1 mixture of diastereomers. When the ¹H NMR spectrum of pyrrolidine **19** was compared to the ¹H NMR spectrum of pyrrolidine 10 prepared via the bicyclic lactam route, it became apparent that the minor diastereomer of 19 was identical to 10. With the absolute configuration of 19 known, this sequence proved the relative stereochemistry of 10, with the acetoxymethyl substituents lying *anti* to each other. Since (+)-2 had previously been prepared in the literature,^{5b} to confirm the absolute configuration of 10, it only became necessary to determine the stereochemistry of 2 prepared from 10 above ($[\alpha]_D$ lit. =



Scheme 3.

+30.0,¹⁴ $[\alpha]_D$ of 2=-44.2). The negative optical rotation observed for 2 proved that the stereochemical assignment was correct, with the dihydroxylation proceeding from the *exo*-face and the C–O ring reduction occurring with retention of configuration to afford (-)-2.

In conclusion, an efficient, enantioselective route to branched, *N*-substituted pyrrolidines has been accomplished using chiral bicyclic lactams. A convenient entry into branched pyrrolidines is now available using intermediate **10**, which may be elaborated in a variety of ways.

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References

- 1. (a) Baxter, E. W.; Reitz, A. B. J. Org. Chem. 1994, 59, 3175. (b) Sinnott, L. M. Chem. Rev. 1990, 90, 1171.
- 2. Asano, N.; Nash, R. J.; Molyneux, R. J.; Fleet, G. W. J. Tetrahedron: Asymmetry 2000, 11, 1645.
- 3. (a) Bernaki, R. J.; Korytnyk, W. Cancer Metastasis Rev. 1985, 4, 81. (b) Tsukamoro, K. et al. Clin. Res. 1989, 37A, 722.
- (a) Fleet, G. W. J. et al; *FEBS Lett.* 1998, 237. (b) Gruters, R. A.; Niefies, J. J.; Tersmette, M. I.; DeGroede, R. E.; Tulp, A.; Heisman, H. G.; Miedema, F.; Ploegh, H. L. *Nature* 1987, 330, 74.
- 5. (a) Ichikawa, M.; Igarashi, Y.; Ichikawa, Y. Tetrahedron Lett. 1995, 36, 1767. (b) Bennis, K.; Gelas, J.; Thomassigny, C. Carbohydr. Res. 1995, 279, 307.
- Meyers, A. I.; Andres, C. J.; Resek, J. E.; Woodall, C. C.; McLaughlin, M. A.; Lee, P. H.; Price, D. A. Tetrahedron 1999, 55, 8931.
- 7. Romo, D.; Meyers, A. I. Tetrahedron 1991, 47, 9503.
- 8. Grieco, P. A. Synthesis 1975, 67 and references cited therein.
- 9. Forns, P.; Fernndez, M. M.; Diez, A.; Rubiralta, M.; Cherrier, M. P.; Bonin, M.; Quirion, J.-C. Synthesis 1999, 258.
- 10. Lee, K.; Wiemer, D. F. J. Org. Chem. 1991, 56, 5556.
- 11. Kolb, H. C.; VanNieuwenhze, M. S.; Sharpless, K. B. Chem. Rev. 1994, 94, 2483.
- 12. Meyers, A. I.; Burgess, L. E. J. Org. Chem. 1991, 56, 2294.
- 13. Czech, B. P.; Bartsch, R. A. J. Org. Chem. 1984, 49, 4076.
- 14. The $[\alpha]_D$ value of our sample, -44.2, does not compare well to the literature value, +30.0, because their compound was contaminated with trace amounts of another undetermined isomer—see Ref. 5b.